

# Appendix VII

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## Some chemical symbols used in this report

<b>C</b>	carbon (there are three isotopes: $^{12}\text{C}$ , $^{13}\text{C}$ , $^{14}\text{C}$ )	<b>DOC</b>	dissolved organic carbon
<b>Ca</b>	calcium	<b>H<sub>2</sub></b>	hydrogen
<b>CaCO<sub>3</sub></b>	calcium carbonate	<b>halon-1211</b>	CF <sub>2</sub> ClBr
<b>CCl<sub>4</sub></b>	carbon tetrachloride	<b>halon-1301</b>	CF <sub>3</sub> Br
<b>CF<sub>4</sub></b>	perfluoromethane	<b>halon-2402</b>	CF <sub>2</sub> BrCF <sub>2</sub> Br
<b>C<sub>2</sub>F<sub>6</sub></b>	perfluoroethane	<b>HCFC</b>	hydrochlorofluorocarbon
<b>C<sub>3</sub>F<sub>8</sub></b>	perfluoropropane	<b>HCFC-21</b>	CHCl <sub>2</sub> F
<b>C<sub>4</sub>F<sub>8</sub></b>	perfluorocyclobutane	<b>HCFC-22</b>	CHF <sub>2</sub> Cl
<b>C<sub>4</sub>F<sub>10</sub></b>	perfluorobutane	<b>HCFC-123</b>	C <sub>2</sub> F <sub>3</sub> HCl <sub>2</sub>
<b>C<sub>5</sub>F<sub>12</sub></b>	perfluoropentane	<b>HCFC-124</b>	CF <sub>3</sub> CHClF
<b>C<sub>6</sub>F<sub>14</sub></b>	perfluorohexane	<b>HCFC-141b</b>	CH <sub>3</sub> CFCl <sub>2</sub>
<b>CFC</b>	chlorofluorocarbon	<b>HCFC-142b</b>	CH <sub>3</sub> CF <sub>2</sub> Cl
<b>CFC-11</b>	CFCl <sub>3</sub> (trichlorofluoromethane)	<b>HCFC-225ca</b>	CF <sub>3</sub> CF <sub>2</sub> CHCl <sub>2</sub>
<b>CFC-12</b>	CF <sub>2</sub> Cl <sub>2</sub> (dichlorodifluoromethane)	<b>HCFC-225cb</b>	CClF <sub>2</sub> CF <sub>2</sub> CHClF
<b>CFC-13</b>	CF <sub>3</sub> Cl (chlorotrifluoromethane)	<b>HCFC-235da2</b>	CF <sub>3</sub> CHClOCHF <sub>2</sub>
<b>CFC-113</b>	CF <sub>2</sub> ClCFCl <sub>2</sub> (trichlorotrifluoroethane)	<b>HCO<sub>3</sub><sup>-</sup></b>	bicarbonate ion
<b>CFC-114</b>	CF <sub>2</sub> ClCF <sub>2</sub> Cl (dichlorotetrafluoroethane)	<b>HFC</b>	hydrofluorocarbon
<b>CFC-115</b>	CF <sub>3</sub> CF <sub>2</sub> Cl (chloropentafluoroethane)	<b>HFC-23</b>	CHF <sub>3</sub>
<b>CF<sub>3</sub>I</b>	trifluoroiodomethane	<b>HFC-32</b>	CH <sub>2</sub> F <sub>2</sub>
<b>CH<sub>4</sub></b>	methane	<b>HFC-41</b>	CH <sub>3</sub> F
<b>C<sub>2</sub>H<sub>6</sub></b>	ethane	<b>HFC-125</b>	CHF <sub>2</sub> CF <sub>3</sub>
<b>C<sub>5</sub>H<sub>8</sub></b>	isoprene	<b>HFC-134</b>	CHF <sub>2</sub> CHF <sub>2</sub>
<b>C<sub>6</sub>H<sub>6</sub></b>	benzene	<b>HFC-134a</b>	CF <sub>3</sub> CH <sub>2</sub> F
<b>C<sub>7</sub>H<sub>8</sub></b>	toluene	<b>HFC-143</b>	CH <sub>2</sub> F CHF <sub>2</sub>
<b>C<sub>10</sub>H<sub>16</sub></b>	terpene	<b>HFC-143a</b>	CH <sub>3</sub> CF <sub>3</sub>
<b>CH<sub>3</sub>Br</b>	methylbromide	<b>HFC-152</b>	CH <sub>2</sub> FCH <sub>2</sub> F
<b>CH<sub>3</sub>CCl<sub>3</sub></b>	methyl chloroform	<b>HFC-152a</b>	CH <sub>3</sub> CHF <sub>2</sub>
<b>CHCl<sub>3</sub></b>	chloroform/trichloromethane	<b>HFC-161</b>	CH <sub>3</sub> CH <sub>2</sub> F
<b>CH<sub>2</sub>Cl<sub>2</sub></b>	dichloromethane/methylene chloride	<b>HFC-227ea</b>	CF <sub>3</sub> CHFCF <sub>3</sub>
<b>CH<sub>3</sub>Cl</b>	methylchloride	<b>HFC-236cb</b>	CF <sub>3</sub> CF <sub>2</sub> CH <sub>2</sub> F
<b>CH<sub>3</sub>OCH<sub>3</sub></b>	dimethyl ether	<b>HFC-236ea</b>	CF <sub>3</sub> CHFCHF <sub>2</sub>
<b>CO</b>	carbon monoxide	<b>HFC-236fa</b>	CF <sub>3</sub> CH <sub>2</sub> CF <sub>3</sub>
<b>CO<sub>2</sub></b>	carbon dioxide	<b>HFC-245ca</b>	CH <sub>2</sub> FCF <sub>2</sub> CHF <sub>2</sub>
<b>CO<sub>3</sub><sup>2-</sup></b>	carbonate ion	<b>HFC-245ea</b>	CHF <sub>2</sub> CHFCHF <sub>2</sub>
<b>DIC</b>	dissolved inorganic carbon	<b>HFC-245eb</b>	CF <sub>3</sub> CHFCH <sub>2</sub> F

<b>HFC-245fa</b>	$\text{CHF}_2\text{CH}_2\text{CF}_3$	<b>HFOC-134</b>	$\text{CF}_2\text{HOCF}_2\text{H}$
<b>HFC-263fb</b>	$\text{CF}_3\text{CH}_2\text{CH}_3$	<b>HFOC-143a</b>	$\text{CF}_3\text{OCH}_3$
<b>HFC-338pcc</b>	$\text{CHF}_2\text{CF}_2\text{CF}_2\text{CF}_2\text{H}$	<b>HFOC-152a</b>	$\text{CH}_3\text{OCHF}_2$
<b>HFC-356mcf</b>	$\text{CF}_3\text{CF}_2\text{CH}_2\text{CH}_2\text{F}$	<b>HFOC-245fa</b>	$\text{CHF}_2\text{OCH}_2\text{CF}_3$
<b>HFC-356mff</b>	$\text{CF}_3\text{CH}_2\text{CH}_2\text{CF}_3$	<b>HFOC-356mmf</b>	$\text{CF}_3\text{CH}_2\text{OCH}_2\text{CF}_3$
<b>HFC-365mfc</b>	$\text{CF}_3\text{CH}_2\text{CF}_2\text{CH}_3$	<b>HG-01</b>	$\text{CHF}_2\text{OCF}_2\text{CF}_2\text{OCHF}_2$
<b>HFC-43-10mee</b>	$\text{CF}_3\text{CHFCHFCF}_2\text{CF}_3$	<b>HG-10</b>	$\text{CHF}_2\text{OCF}_2\text{OCHF}_2$
<b>HFC-458mfcf</b>	$\text{CF}_3\text{CH}_2\text{CF}_2\text{CH}_2\text{CF}_3$	<b>H-Galden 1040x</b>	$\text{CHF}_2\text{OCF}_2\text{OC}_2\text{F}_4\text{OCHF}_2$
<b>HFC-55-10mccff</b>	$\text{CF}_3\text{CF}_2\text{CH}_2\text{CH}_2\text{CF}_2\text{CF}_3$	<b>HNO<sub>3</sub></b>	nitric acid
<b>HFE-125</b>	$\text{CF}_3\text{OCHF}_2$	<b>HO<sub>2</sub></b>	hydroperoxyl
<b>HFE-134</b>	$\text{CF}_2\text{HOCF}_2\text{H}$	<b>HO<sub>x</sub></b>	the sum of OH and HO <sub>2</sub>
<b>HFE-143a</b>	$\text{CF}_3\text{OCH}_3$	<b>H<sub>2</sub>O</b>	water vapour
<b>HFE-152a</b>	$\text{CH}_3\text{OCHF}_2$	<b>H<sub>2</sub>SO<sub>4</sub></b>	sulphuric acid
<b>HFE-227ea</b>	$\text{CF}_3\text{CHFOCF}_3$	<b>N<sub>2</sub></b>	molecular nitrogen
<b>HFE-236ea2</b>	$\text{CF}_3\text{CHFOCHF}_2$	<b>NF<sub>3</sub></b>	nitrogen trifluoride
<b>HFE-236fa</b>	$\text{CF}_3\text{CH}_2\text{OCF}_3$	<b>NH<sub>3</sub></b>	ammonia
<b>HFE-245cb2</b>	$\text{CF}_3\text{CF}_2\text{OCH}_3$	<b>NH<sub>4</sub><sup>+</sup></b>	ammonium ion
<b>HFE-245fa1</b>	$\text{CHF}_2\text{CH}_2\text{OCF}_3$	<b>NMHC</b>	non-methane hydrocarbon
<b>HFE-245fa2</b>	$\text{CHF}_2\text{OCH}_2\text{CF}_3$	<b>NO</b>	nitric oxide
<b>HFE-254cb2</b>	$\text{CHF}_2\text{CF}_2\text{OCH}_3$	<b>NO<sub>2</sub></b>	nitrogen dioxide
<b>HFE-263fb2</b>	$\text{CF}_3\text{CH}_2\text{OCH}_3$	<b>NO<sub>x</sub></b>	nitrogen oxides (the sum of NO and NO <sub>2</sub> )
<b>HFE-329mcc2</b>	$\text{CF}_3\text{CF}_2\text{OCF}_2\text{CHF}_2$	<b>NO<sub>3</sub></b>	nitrate radical
<b>HFE-338mcf2</b>	$\text{CF}_3\text{CF}_2\text{OCH}_2\text{CF}_3$	<b>NO<sub>3</sub><sup>-</sup></b>	nitrate ion
<b>HFE-347mcc3</b>	$\text{CF}_3\text{CF}_2\text{CF}_2\text{OCH}_3$	<b>N<sub>2</sub>O</b>	nitrous oxide
<b>HFE-347mcf2</b>	$\text{CF}_3\text{CF}_2\text{OCH}_2\text{CHF}_2$	<b>O<sub>2</sub></b>	molecular oxygen
<b>HFE-356mcc3</b>	$\text{CF}_3\text{CHF}_2\text{OCH}_3$	<b>O<sub>3</sub></b>	ozone
<b>HFE-356mff2</b>	$\text{CF}_3\text{CH}_2\text{OCH}_2\text{CF}_3$	<b>OCS</b>	organic carbonyl sulphide
<b>HFE-356pcc3</b>	$\text{CHF}_2\text{CF}_2\text{CF}_2\text{OCH}_3$	<b>OH</b>	hydroxyl radical
<b>HFE-356pcf2</b>	$\text{CHF}_2\text{CF}_2\text{OCH}_2\text{CHF}_2$	<b>PAN</b>	peroxyacetyl nitrate
<b>HFE-356pcf3</b>	$\text{CHF}_2\text{CF}_2\text{CH}_2\text{OCHF}_2$	<b>PFC</b>	perfluorocarbon
<b>HFE-365mcf3</b>	$\text{CF}_3\text{CF}_2\text{CH}_2\text{OCH}_3$	<b>SF<sub>6</sub></b>	sulphur hexafluoride
<b>HFE-374pc2</b>	$\text{CHF}_2\text{CF}_2\text{OCH}_2\text{CH}_3$	<b>SF<sub>5</sub>CF<sub>3</sub></b>	trifluoromethyl sulphur pentafluoride
<b>HFE-7100</b>	$\text{C}_4\text{F}_9\text{OCH}_3$	<b>SO<sub>2</sub></b>	sulphur dioxide
<b>HFE-7200</b>	$\text{C}_4\text{F}_9\text{OC}_2\text{H}_5$	<b>SO<sub>4</sub><sup>2-</sup></b>	sulphate ion
<b>HFOC-125</b>	$\text{CF}_3\text{OCHF}_2$	<b>VOC</b>	volatile organic compounds